

Name _____ Date _____ Class _____

16**SOLUTIONS****SECTION 16.1 PROPERTIES OF SOLUTIONS (pages 471–477)**

This section identifies the factors that affect the solubility of a substance and determine the rate at which a solute dissolves.

► Solution Formation (pages 471–472)

Look at Figure 16.1 on page 471 to help you answer Questions 1 and 2.

- Underline the condition that causes sugar to dissolve faster in water.
 - as a whole cube or in granulated form?
 - when allowed to stand or when stirred?
 - at a higher temperature or a lower temperature?
- Name three factors that influence the rate at which a solute dissolves in a solvent.
 - agitation (stirring or shaking)
 - temperature
 - particle size
- Is the following sentence true or false? Finely ground particles dissolve more rapidly than larger particles because finer particles expose a greater surface area to the colliding solvent molecules. true

► Solubility (pages 472–473)

- Complete the following table showing the steps in a procedure to determine the total amount of sodium chloride that will dissolve in 100 g of water at 25°C.

Procedure	Amount Dissolved	Amount Not Dissolved
Add 36.0 g of sodium chloride to the water	36.0 g	0.0 g
Add an additional 1.0 g of sodium chloride	<u>0.2 g</u>	<u>0.8 g</u>
Determine the total amount that dissolves	<u>36.2 g</u>	

- The amount of a substance that dissolves in a given quantity of solvent at a constant temperature is called the substance's solubility at that temperature.
- If a solution contains the maximum amount of solute for a given quantity of solvent at a constant temperature, it is called a(n) saturated solution.

Properties Of Solutions Answers

Ilia Manolov



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Foundations of College Chemistry Morris Hein, Susan Arena, 2013-01-01 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years Foundations of College Chemistry Alternate 14th Edition has helped readers master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis **E3 Chemistry Review Book - 2018 Home Edition (Answer Key Included)** Effiong Eyo, 2017-10-20 With Answer Key to All Questions Chemistry students and homeschoolers Go beyond just passing Enhance your understanding of chemistry and get higher marks on homework quizzes tests and the regents exam with E3 Chemistry Review Book 2018 With E3 Chemistry Review Book students will get clean clear engaging exciting and easy to understand high school chemistry concepts with emphasis on New York State Regents Chemistry the Physical Setting Easy to read format to help students easily remember key and must know chemistry materials Several example problems with solutions to study and follow Several practice multiple choice and short answer questions at the end of each lesson to test understanding of the materials 12 topics of Regents question sets and 3 most recent Regents exams to practice and prep for any Regents Exam This is the Home Edition of the book Also available in School Edition ISBN 978 197836229 The Home Edition contains an answer key section Teachers who want to recommend our Review Book to their students should recommend the Home Edition Students and and parents whose school is not using the Review Book as instructional material as well as homeschoolers should buy the Home Edition The School Edition does not have answer key in the book A separate answer key booklet is provided to teachers with a class order of the book Whether you are using the school or Home Edition our E3 Chemistry Review Book makes a great supplemental instructional and test prep resource that can be used from the beginning to the end of the school year PLEASE NOTE Although reading contents in both the school and home editions are identical there are slight differences in question numbers choices and pages between the two editions Students whose school is using the Review Book as instructional material SHOULD NOT buy the Home Edition Also available in paperback print *CBSE Chapterwise Instant Notes Class 12 Chemistry Book* MTG Learning Media, MTG presents a new resource to help CBSE board students with this masterpiece Chapterwise Instant Notes This book is the best revision resource for CBSE students as it has instant chapter wise notes for complete latest CBSE syllabus The book comprises chapter wise quick recap notes and then a lot of subjective questions which covers the whole chapter in the form of these questions OAR Quarterly Index of Current Research Results United States. Air Force. Office of Aerospace Research, 1967 *Intelligent Automation and Systems Engineering* Sio-Long Ao, Mahyar Amouzegar, Burghard B. Rieger, 2011-08-23 Intelligent systems are required to facilitate the use of information provided by the internet and other

computer based technologies This book describes the state of the art in Intelligent Automation and Systems Engineering Topics covered include Intelligent decision making Automation Robotics Expert systems Fuzzy systems Knowledge based systems Knowledge extraction Large database management Data analysis tools Computational biology Optimization algorithms Experimental designs Complex system identification Computational modeling Systems simulation Decision modeling and industrial applications

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