## **Chapter 3 Study Guide**

electrons.

#### SECTION 3.1 Matter

automs.

In your textbook, read about elements and atomic structure.

element.

Use each of the terms below just once to complete the passage.

A(n) (1)		is a substance that cannot be broken down into simpler substances.		
AJ()	n) (2)	is the smallest particle of matter having all that element's		
ch	characteristics. It is made up of smaller particles. The (3)is			
ma	de up of protons and neutron	s. Small particles that have mass and posi	tive electrical charges	
ane	(4)	_ Particles that have about the same mass	les that have about the same mass as protons, but that	
are electrically neutral are (5)		Surrounding the	Surrounding the nucleus of an atom are	
tiny particles called (6)		, which have little mass,	, which have little mass, but have negative	
ele	ctrical charges that are exact	y the same magnitude as the positive char	ges of protons.	
	textbook, read about atomic te each statement.	structure and isotopes.		
7.	The number of protons in an atom's nucleus is the			
3. When atoms of the same element have different mass numbers, they are kn			are known	
	as	of that element		
9.	The spontaneous process through which unstable nuclei emit radiation is called			
10	A(n) represents the area in an atom where an electron is		m electron is	
Admin N	most likely to be found.		nau-rotte chu an	
11	An atom that gains or loses an electron and has an electric charge is called a(n)			
1.2	The combined number of protons and neutrons is the			
1.3.	The	is the average of the mass numbers of t	he isotopes of an	
	element.			

meutrons:

marchesis.

persons.

# **Section 3 Study Guide Answer Key**

**Henry Davis** 

#### **Section 3 Study Guide Answer Key:**

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