

PROBLEM SET 3: EQUILIBRIUM PROBLEMS

1. Write the expression for the equilibrium constant for this reaction.



2. Calculate the equilibrium constant for the reaction in Problem 1 if the equilibrium concentrations are: $[\text{N}_2\text{O}_5] = 0.50 \text{ mol/L}$, $[\text{NO}_2] = 0.80 \text{ mol/L}$, $[\text{O}_2] = 0.20 \text{ mol/L}$.
3. How would the equilibrium position for the equation in Problem 1 be effected by
 - a. an addition of O_2 to the reaction vessel?
 - b. a decrease in the pressure?
4. The equilibrium constant for the dimerization of nitrogen dioxide to form dinitrogen tetroxide is 5.6.



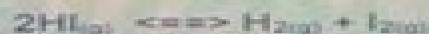
In a two-liter container the amount of N_2O_4 , at equilibrium, is 0.66 mol. What is the equilibrium concentration of NO_2 ?

5. What effect would an increase in pressure have on the equilibrium position of the reaction in Problem 4?
6. Which value of K_{eq} indicates most favorability of product formation, $K_{eq} = 1 \times 10^{-12}$, $K_{eq} = 1.5$, or $K_{eq} = 5.6 \times 10^{-4}$? (ie, which K_{eq} suggests higher conc. of products than reactants?)
7. Hydrogen sulfide gas decomposes into its elements and establishes an equilibrium at 1400C.



A liter of this gas mixture contains 0.18 mol H_2S , 0.014 mol H_2 and 0.035 mol S_2 . Calculate the equilibrium constant, K_{eq} , for this reaction.

8. A 2.00L flask was filled with 4.00 mol of HI at a certain temperature and given sufficient time to react. At equilibrium the concentration of H_2 was 0.400 M. Find the equilibrium concentrations of I_2 and HI and then find the K_{eq} at this temperature.



Reaction Rate Equilibrium Word Problems

Michel Boudart



Reaction Rate Equilibrium Word Problems:

How to Solve Word Problems in Chemistry David E Goldberg, 2001-06-26 In addition to having to master a vast number of difficult concepts and lab procedures high school chemistry students must also learn with little or no coaching from their teachers how to solve tough word problems Picking up where standard chemistry texts leave off *How to Solve Word Problems in Chemistry* takes the fear and frustration out of chemistry word problems by providing students with easy to follow procedures for solving problems in everything from radioactive half life to oxidation reduction reactions

Reaction Kinetics Based on Time-Energy Uncertainty Principle Satoru Yamamoto, 2023-05-26 This book proposes a completely unique reaction kinetics theory based on the uncertainty principle of quantum mechanics the physical viewpoint and mathematical details for the theory construction are explained and abundant applications of the theory mainly in materials science are described The theory argues that physical systems on reaction are in a quantum mechanically uncertain state and that such systems will transition to new states after a finite duration time Based on this theory if the magnitude of the energy uncertainty i e energy fluctuation of the system on reaction can be determined we can calculate the reaction rates not only for the thermal activation processes but also for the non thermal activation process such as mechanical optical electromagnetic or other actions Therefore researchers or engineers who are involved in fields such as the discovery of new chemical substances development of materials innovation of manufacturing processes and also everyone purely interested in kinetic methodology find this book very stimulating and motivating

Materials Kinetics Fundamentals Ryan O'Hayre, 2015-01-12 Introductory kinetics for the undergrad materials scientist *Materials Kinetics Fundamentals* is an accessible and interesting introduction to kinetics processes with a focus on materials systems Designed for the undergraduate student this book avoids intense mathematics to present the theory and application of kinetics in a clear reader friendly way Students are first introduced to the fundamental concepts of kinetics with illustrated diagrams examples text boxes and homework questions that impart a unified intuitive understanding Further chapters cover the application of these concepts in the context of materials science with real world examples including silicon processing and integrated circuit fabrication thin film deposition carbon 14 dating steel degassing energy conversion and more Instructor materials including a test bank are available through the companion website providing a complete resource for the undergraduate materials science student At its core kinetics deals with rates telling us how fast something will take place for example how fast water will evaporate or how fast molten silicon will solidify This book is designed to provide students with an introduction to kinetics underlying principles without rigorous math to distract from understanding Understand universally important kinetic concepts like diffusion and reaction rate Model common kinetic processes both quantitatively and qualitatively Learn the mechanisms behind important and interesting materials systems Examine the behaviors properties and interactions of relevant solid materials There are a large number of books on chemical kinetics but there are far fewer

that focus on materials kinetics and virtually none that provide an accessible introductory level treatment of the subject Materials Kinetics Fundamentals fills that need with clear detailed explanations of these universal concepts

Computational Subsurface Hydrology Gour-Tsyh (George) Yeh, 2012-12-06 Any numerical subsurface model is comprised of three components a theoretical basis to translate our understanding phenomena into partial differential equations and boundary conditions a numerical method to approximate these governing equations and implement the boundary conditions and a computer implementation to generate a generic code for research as well as for practical applications Computational Subsurface Hydrology Reactions Transport and Fate is organized around these themes The fundamental processes occurring in subsurface media are rigorously integrated into governing equations using the Reynolds transport theorem and interactions of these processes with the surrounding media are sophisticatedly cast into various types of boundary conditions using physical reasoning A variety of numerical methods to deal with reactive chemical transport are covered in Computational Subsurface Hydrology Reactions Transport and Fate with a particular emphasis on the adaptive local grid refinement and peak capture using the Lagrangian Eulerian approach The topics on coupled fluid flows and reactive chemical transport are unique contributions of this book They serve as a reference for research as well as for practical applications with a computer code that can be purchased from the author Four computer codes to simulate vertically integrated horizontal solute transport LEMA contaminant transport in moving phreatic aquifers in three dimensions 3DLEMA solute transport in variably saturated flows in two dimensions LEWASTE and solute transport under variably saturated flows in three dimensions 3DLEWASTE are covered These four computer codes are designed for generic applications to both research and practical problems They could be used to simulate most of the practical real world field problems Reactive chemical transport and its coupling with fluid flows are unique features in this book Theories numerical implementations and example problems of coupled reactive transport and flows in variably saturated media are presented A generic computer code HYDROGEOCHEM 3.0 is developed A total of eight example problems are used to illustrate the application of the computational model These problems are intended to serve as examples for setting up a variety of simulations that one may encounter in research and field site applications Computational Subsurface Hydrology Reactions Transport and Fate offers practicing engineers and scientists a theoretical background numerical methods and computer codes for modeling contaminant transport in subsurface media It also serves as a textbook for senior and graduate course on reactive chemical transport in subsurface media in disciplines such as civil and environmental engineering agricultural engineering geosciences soil sciences and chemical engineering Computational Subsurface Hydrology Reactions Transport and Fate presents a systematic derivation of governing equations and boundary conditions of subsurface contaminant transport as well as reaction based geochemical and biochemical processes It discusses a variety of numerical methods for moving sharp front problems expounds detail procedures of constructing Lagrangian Eulerian finite element methods and describes precise

implementation of computer codes as they are applied to subsurface contaminant transport and biogeochemical reactions

A Problem-Solving Approach to Aquatic Chemistry James N. Jensen, 2023-01-05 A Problem Solving Approach to Aquatic Chemistry Enables civil and environmental engineers to understand the theory and application of aquatic equilibrium chemistry The second edition of A Problem Solving Approach to Aquatic Chemistry provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics The text directly addresses two required ABET program outcomes in environmental engineering chemistry including stoichiometry equilibrium and kinetics and material and energy balances fate and transport of substances in and between air water and soil phases The book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter's main points To aid in reader comprehension important terms are defined in context and key ideas are summarized Many thought provoking discussion questions worked examples and end of chapter problems are also included Each part of the text begins with a case study a portion of which is addressed in each subsequent chapter illustrating the principles of that chapter In addition each chapter has an Historical Note exploring connections with the people and cultures connected to topics in the text A Problem Solving Approach to Aquatic Chemistry includes Fundamental concepts such as concentration units thermodynamic basis of equilibrium and manipulating equilibria Solutions of chemical equilibrium problems including setting up the problems and algebraic graphical and computer solution techniques Acid base equilibria including the concepts of acids and bases titrations and alkalinity and acidity Complexation including metals ligands equilibrium calculations with complexes and applications of complexation chemistry Oxidation reduction equilibria including equilibrium calculations graphical approaches and applications Gas liquid and solid liquid equilibrium with expanded coverage of the effects of global climate change Other topics including chemical kinetics of aquatic systems surface chemistry and integrative case studies For advanced senior undergraduates and first year graduate students in environmental engineering courses A Problem Solving Approach to Aquatic Chemistry serves as an invaluable learning resource on the topic with a variety of helpful learning elements included throughout to ensure information retention and the ability to apply covered concepts in practical settings

Kinetics of Chemical Processes Michel Boudart, 2014-05-16

Kinetics of Chemical Processes details the concepts associated with the kinetic study of the chemical processes The book is comprised of 10 chapters that present information relevant to applied research The text first covers the elementary chemical kinetics of elementary steps and then proceeds to discussing catalysis The next chapter tackles simplified kinetics of sequences at the steady state Chapter 5 deals with coupled sequences in reaction networks while Chapter 6 talks about autocatalysis and inhibition The seventh chapter describes the irreducible transport phenomena in chemical kinetics The next two chapters discuss the correlations in homogenous kinetics and heterogeneous catalysis respectively The last chapter covers the analysis of reaction networks The book will be of great use to students researchers and practitioners of scientific

disciplines that deal with chemical reaction particularly chemistry and chemical engineering *PCAT Test Prep Biology Review--Exambusters Flash Cards--Workbook 3 of 4* PCAT Exambusters,2016-06-01 PCAT Prep Flashcard Workbook 3 BIOLOGY REVIEW 450 questions and answers ILLUSTRATED Essential definitions and concepts Topics Cells Biochemistry and Energy Evolution and Classification Kingdoms Bacteria Fungi Protista Kingdom Plantae Kingdom Animalia Human Locomotion Human Circulation and Immunology Human Respiration and Excretion Human Digestion Human Nervous System Human Endocrinology Reproduction and Development Genetics Ecology ADDITIONAL WORKBOOKS PCAT Prep Flashcard Workbook 4 CHEMISTRY REVIEW 700 questions and answers Essential definitions formulas concepts and sample problems Topics Introduction Matter Atoms Formulas Moles Reactions Elements Periodic Table Electrons Chemical Bonds Heat Gases Phase Changes Solutions Reaction Rates Equilibrium Acids and Bases Oxidation and Reduction Introduction to Organic Chemistry Radioactivity _____ PCAT Prep Flashcard Workbook 2 ALGEBRA REVIEW 450 questions and answers that highlight introductory algebra definitions problems and concepts Topics Algebraic Concepts Sets Variables Exponents Properties of Numbers Simple Equations Signed Numbers Monomials Polynomials Additive and Multiplicative Inverse Word Problems Prime Numbers Factoring Algebraic Fractions Ratio and Proportion Variation Radicals Quadratic Equations Exambusters PCAT Prep Workbooks provide comprehensive fundamental PCAT review one fact at a time to prepare students to take practice PCAT tests Each PCAT study guide focuses on one specific subject area covered on the PCAT exams From 300 to 600 questions and answers each volume in the PCAT series is a quick and easy focused read Reviewing PCAT flash cards is the first step toward more confident PCAT preparation and ultimately higher PCAT exam scores Kinetics of Chemical Reactions Guy B. Marin,Gregory S. Yablonsky,Denis Constales,2019-04-29 This second extended and updated edition presents the current state of kinetics of chemical reactions combining basic knowledge with results recently obtained at the frontier of science Special attention is paid to the problem of the chemical reaction complexity with theoretical and methodological concepts illustrated throughout by numerous examples taken from heterogeneous catalysis combustion and enzyme processes Of great interest to graduate students in both chemistry and chemical engineering Air Pollution Abstracts ,1974 *Computational Fluid Dynamics Review 1998 (In 2 Volumes)* Mohamed M Hafez,Koichhi Oshima,1998-11-20 The first volume of CFD Review was published in 1995 The purpose of this new publication is to present comprehensive surveys and review articles which provide up to date information about recent progress in computational fluid dynamics on a regular basis Because of the multidisciplinary nature of CFD it is difficult to cope with all the important developments in related areas There are at least ten regular international conferences dealing with different aspects of CFD It is a real challenge to keep up with all these activities and to be aware of essential and fundamental contributions in these areas It is hoped that CFD Review will help in this regard by covering the state of the art in this field The present book contains sixty two articles written by authors from the US Europe Japan and China covering the

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