



Reaction Energy



Tools

REACTION INVESTIGATION

Starting temperature: 21.0 °C

CONTROL

Reaction 1

$$2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$$

☒ Forward ☐ Reverse

Key

H ₂		H ₂ O	
O ₂			

Instructions

Click **Play** to start the reaction. The reaction chamber contains 0.2 moles of H₂ and 0.1 moles of O₂.

Theoretical ΔH : — kJ

Experimental ΔH : — kJ



ASSESSMENT QUESTIONS

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Questions & Answers

1. A reaction occurs in a calorimeter, resulting in the starting and final temperatures shown below. What can you say about the reaction and the enthalpy change (ΔH) during the reaction?

Reaction Energy Study Guide Answers

Brian W. Stewart



Reaction Energy Study Guide Answers:

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