

MIXTURES & SOLUTIONS

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Study Guide

MIXTURES & SOLUTIONS Study Guide

Before each of the following activities, I can identify the purpose of the activity. I can describe the materials that are needed for the activity. I can explain how to separate mixtures and which method to use.

Word	Definition
Mixture	
Solution	
Dissolve	
Solvent	
Solute	
Substance	

- Highlight any phrases that are describing a solution.
- Excludes only solid substances
 - Excludes liquids
 - Excludes substances in solid
 - Must be the same substance
 - Must be the same substance
 - Something dissolves

Order the mixtures below:



Do you want a substance to dissolve quicker?

- Decrease the _____ of the solvent
- _____ the substance in the solvent
- Put the solute into the container _____
- the surface area

Name 1 substance that dissolves in water:

Name 1 substance that does NOT dissolve in water:

What can you do to help a substance dissolve faster?

SEPARATING MIXTURES

Required materials	Describe what happens
Method name: Decanting Objects of different densities	Place objects of different densities into water. Observe objects and note which objects float.
Method name: Pipette Mixture (liquid) separated into layers	Separating liquids from each other using a pipette.
Method name: Filter Solid particles Liquid Filter	Separate solid particles from liquids by pouring them through a filter. Observe what is left on the filter.
Method name: Evaporation Solid particles Liquid particles	Separate solid particles from liquids by pouring them through a filter. Observe what is left on the filter.
Method name: Distillation Liquid particles	Separate a liquid from a mixture by heating it until it boils. Observe what is left on the filter.

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