STOICHIOMETRY: MIXED PROBLEMS

Name _____

N₂ + 3H₂ → 2NH₃

What volume of NH $_{\rm 3}$ at STP is produced if 25.0 g of N $_{\rm 2}$ is reacted with an excess of H $_{\rm 2}$

2KCIO₃ → 2KCI + 3O₂

If 5.0 g of KClO $_3$ is decomposed, what volume of O $_2$ is produced at STP?

- 3. How many grams of KCI are produced in Problem 2?
- 4. Zn + 2HCl → ZnCl, + H,

What volume of hydrogen at STP is produced when 2.5 g of zinc react with an excess of hydrochloric acid?

5. H,SO₄ + 2NaOH → H₂O + Na₃SO₄

How many molecules of water are produced if 2.0 g of sodium sulfate are produced in the above reaction?

6. 2AICI, → 2AI + 3CI,

If 10.0 g of aluminum chloride are decomposed, how many molecules of Cl₂ are produced?

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Peipei Pang

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