

NUCLEAR CHEMISTRY WORKSHEET

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Most nuclear reactions involve changes in the number of protons and/or neutrons in the nucleus of an atom. These changes are called *transmutations* because an atom of one element is changed into an atom of a different element. During these changes, nuclear particles may be absorbed and/or emitted along with a release of energy.

There are four types of nuclear reactions:

- 1) *natural radioactive decay*: transmutations occur by natural radioactivity (the ability of a nucleus to emit a nuclear particle and energy without external stimulation);
- 2) *artificial transmutation*: transmutations occur by nuclear disintegration caused by external stimulation as a scientist bombards a nucleus with a particle (the addition of another nuclear particle makes the nucleus unstable);
- 3) *fission*: certain nuclei having a large mass are bombarded with special particles that cause the nuclei to split into two nuclei each having a smaller mass;
- 4) *fusion*: nuclei of light elements are combined to form heavier nuclei.

Nuclear reactions are written using symbols in the A_ZX notation. The symbols for the major nuclear particles involved in nuclear reactions are given below.

Alpha particle	${}^4_2\text{He}$ (an alpha particle is a helium nucleus)
Proton	${}^1_1\text{H}$ (the most common hydrogen nucleus is a proton)
Neutron	${}^1_0\text{n}$
Electron	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$ (also called a beta particle)
Positron	${}^0_1\text{e}$ or ${}^0_1\beta$
Gamma ray	${}^0_0\gamma$

The table below contains information about nuclear reactions involving the emission of radiation.

Type of Reaction	NUCLEAR CHANGES		
	Radiation (Particle)	Effect on the Atomic Number	Effect on the Atomic Mass
Alpha emission (α)	${}^4_2\text{He}$	decrease by 2	decrease by 4
Beta emission (β)	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$	increase by 1	no change
Positron emission (β^+)	${}^0_1\text{e}$ or ${}^0_1\beta$	decrease by 1	no change
Gamma emission	${}^0_0\gamma$	no change	no change
Electron capture (EC)	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$	decrease by 1	no change

Modern Chemistry Nuclear Chemistry Review Answers

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Handbook of Nuclear Chemistry Attila Vértes, Sándor Nagy, Zoltán Klencsár, Rezso György Lovas, Frank Rösch, 2010-12-10

This revised and extended 6 volume handbook set is the most comprehensive and voluminous reference work of its kind in the field of nuclear chemistry. The Handbook set covers all of the chemical aspects of nuclear science starting from the physical basics and including such diverse areas as the chemistry of transactinides and exotic atoms as well as radioactive waste management and radiopharmaceutical chemistry relevant to nuclear medicine. The nuclear methods of the investigation of chemical structure also receive ample space and attention. The international team of authors consists of scores of world renowned experts: nuclear chemists, radiopharmaceutical chemists and physicists from Europe, USA and Asia. The Handbook set is an invaluable reference for nuclear scientists, biologists, chemists, physicists, physicians practicing nuclear medicine, graduate students and teachers virtually all who are involved in the chemical and radiopharmaceutical aspects of nuclear science. The Handbook set also provides further reading via the rich selection of references. Energy Research Abstracts, 1991 **Review** Oak Ridge National Laboratory, 1985 *Nuclear Science Abstracts*, 1976-05

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

Bibliography of Mass Spectroscopy Literature for 1971, 1973 ERDA Energy Research Abstracts United States. Energy Research and Development Administration, 1977 **Dictionary of Occupational Titles**, 1987 **Brief Review for New York** Patrick Kavanagh, 2004 **ERDA Energy Research Abstracts** United States. Energy Research and Development Administration. Technical Information Center, 1976

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Table of Contents Modern Chemistry Nuclear Chemistry Review Answers

1. Understanding the eBook Modern Chemistry Nuclear Chemistry Review Answers
 - The Rise of Digital Reading Modern Chemistry Nuclear Chemistry Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Nuclear Chemistry Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Nuclear Chemistry Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Nuclear Chemistry Review Answers
 - Personalized Recommendations
 - Modern Chemistry Nuclear Chemistry Review Answers User Reviews and Ratings
 - Modern Chemistry Nuclear Chemistry Review Answers and Bestseller Lists
5. Accessing Modern Chemistry Nuclear Chemistry Review Answers Free and Paid eBooks
 - Modern Chemistry Nuclear Chemistry Review Answers Public Domain eBooks
 - Modern Chemistry Nuclear Chemistry Review Answers eBook Subscription Services
 - Modern Chemistry Nuclear Chemistry Review Answers Budget-Friendly Options

6. Navigating Modern Chemistry Nuclear Chemistry Review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Modern Chemistry Nuclear Chemistry Review Answers Compatibility with Devices
 - Modern Chemistry Nuclear Chemistry Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modern Chemistry Nuclear Chemistry Review Answers
 - Highlighting and Note-Taking Modern Chemistry Nuclear Chemistry Review Answers
 - Interactive Elements Modern Chemistry Nuclear Chemistry Review Answers
8. Staying Engaged with Modern Chemistry Nuclear Chemistry Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Chemistry Nuclear Chemistry Review Answers
9. Balancing eBooks and Physical Books Modern Chemistry Nuclear Chemistry Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modern Chemistry Nuclear Chemistry Review Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modern Chemistry Nuclear Chemistry Review Answers
 - Setting Reading Goals Modern Chemistry Nuclear Chemistry Review Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modern Chemistry Nuclear Chemistry Review Answers
 - Fact-Checking eBook Content of Modern Chemistry Nuclear Chemistry Review Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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