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Chemistry – Unit 10 Worksheet 3

1. Element X has two natural isotopes. The isotope with a mass number of 6 has a relative abundance of 7.5%. The isotope with a mass number of 7 has a relative abundance of 92.5%. Determine the average molar mass for the element from these figures. What is the true identity and atomic number of element X?

 The element copper is found to contain the naturally occurring isotopes 29Cu⁶³ and 29Cu⁶⁵. The relative abundances are 69.1% and 30.9% respectively. Calculate the average molar mass of copper.

Uranium has three isotopes with the following percent abundances: 92U²³⁴ (0.0058%), 92U²³⁵(0.71%), 92U²³⁸ (99.23%). What do you expect the molar mass of uranium to be in whole numbers? Why?

4. A sample of silver as it occurs in nature is 52.0% of isotope 47Ag¹⁰⁷ and 48.0% of isotope 47Ag¹⁰⁹. What is the average molar mass of silver? (Compare your result with the value given in the periodic table).

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Spatial Analysis and Modelling Mr. Rohit Manglik, 2024-06-24 Covers spatial analysis techniques and modeling using GIS and statistical tools for applications in geography urban planning and environmental studies **Environmental Organic Chemistry** Rene P. Schwarzenbach, Philip M. Gschwend, Dieter M. Imboden, 2016-11-07 Examines in a pedagogical way all pertinent molecular and macroscopic processes that govern the distribution and fate of organic chemicals in the environment and provides simple modeling tools to quantitatively describe these processes and their interplay in a given environmental system Treats fundamental aspects of chemistry physics and mathematical modeling as applied to environmentally relevant problems and gives a state of the art account of the field Teaches the reader how to relate the structure of a given chemical to its physical chemical properties and intrinsic reactivities Provides a holistic and teachable treatment of phase partitioning and transformation processes as well as a more focused and tailor made presentation of physical mathematical and modeling aspects that apply to environmental situations of concern Includes a large number of questions and problems allowing teachers to explore the depth of understanding of their students or allowing individuals who use the book for self study to check their progress Provides a companion website which includes solutions for all problems as well as a large compilation of physical constants and compound properties Advances in Organometallic Chemistry Anthony F. Hill, Mark J. Fink, 2011-12-12 Almost all branches of chemistry and material science now interface with organometallic chemistry the study of compounds containing carbon metal bonds This widely acclaimed serial contains authoritative reviews that address all aspects of organometallic chemistry a field that has expanded enormously since the publication of Volume 1 in 1964 Fully updated and expanded to reflect recent advances Illustrated with pertinent examples from recent literature Contributions from leading authorities and industry experts **Literature 1975, Part 2** Siegfried Böhme, Ute Esser, Walter Fricke, Ulrich Güntzel-Lingner, Inge Heinrich, Frieda Henn, Dietlinde Krahn, Lutz Schmadel, Hans Scholl, Gert Zech, 2013-03-14 Bihar Board Solved Model Paper Chemistry Class 12 Latest Edition 2026 SBPD Editorial, 2025-10-16 1 Solid State 2 Solution 3 Electro Chemistry 4 Chemical Kinetics 5 Surface Chemistry 6 General

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