

Due the day of the unit test

Chapter 3 Study Guide

Directions: Please answer the following questions/ complete the following statements using the information you learned in chapter 3. Please be sure to construct COMPLETE answers in complete sentences. Partial answers will not be awarded full credit.

Objective 1: Define federalism and explain why it is important to American government and politics.

1. Define the three ways of organizing a nation that were discussed in the text.
 - 1.
 - 2.
 - 3.
2. In what ways does federalism decentralize politics and policies?

Objective 2: Describe how the Constitution divides power between the national and state governments and understand why the supremacy of the national government is the central principle of American federalism.

1. List the three items that are considered the supreme law of the land.
 - 1.
 - 2.
 - 3.
2. What is the significance of the Tenth Amendment?
3. Explain the difference between enumerated powers and implied powers.

Objective 3: Explain the nature of the states' obligations to each other.

1. Name and describe the three general obligations that each state has to every other state under the Constitution.
 - 1.
 - 2.

Prentice Hall Ap Government Study Guide

George Peacock



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web in today's experiment you will determine the solubility product ksp of calcium hydroxide Ca(OH)_2 by measuring the concentration of Ca(OH)_2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of Ca(OH)_2 is represented in equation form as shown below $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$ the solubility product expression describes in mathematical terms the equilibrium

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$ $5 \times 10^{-2} \text{ M}$ calcium ions $5 \times 10^{-2} \text{ M}$ $2 \times 10^{-2} \text{ M}$ hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web the chemical equation used was $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$ the solubility product was determined to be ksp $\text{Ca}^{2+} \text{OH}^-$ from the

titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as $\text{Ca}_3(\text{PO}_4)_2$ is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore $K_{\text{sp}} = [\text{Ca}^{2+}]^3 [\text{PO}_4^{3-}]^2$

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web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest ksp value to the theoretical ksp for $\text{Ca}(\text{OH})_2$ was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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