1. The basic premise of the collision theory is that particles must first

if a reaction is to occur. Additionally, that

with the _____and th

proper amount of ______

2. ______ is the study of reaction rates which

is the note at which the ______ disappear and the products

3. There are five factors affecting the reaction rate:

a. _____ af the _____

b. _____ of the _____

c.

d. ____

e._____

4. Use the collision theory to explain:

A mixture of hydrogen gas and chilorine gas reacts faster when the

volume they occupy is decreased ______

b. Tron filings rust fester than iron wire:

An increase in temperature increases the reaction rate; ______

-

5. Why do you think that industrial chemists are concerned with reaction rates?

Review Sheet Unit 12

Jack C. Richards

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