

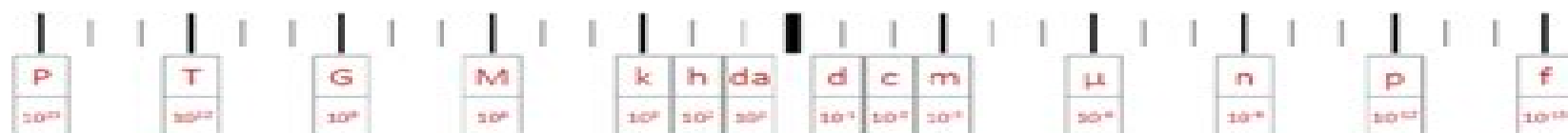
IB Physics SL - Study Guide

Science Skills

List the seven fundamental base units and their abbreviations:

	Unit	Abbreviation
Length	Meter	m
Mass	Kilogram	kg
Time	Second	s
Electric Current	Ampere	A
Temperature	Kelvin	K
Amount of Substance	Mole	mol
Luminous Intensity	Candela	cd

Metric Prefixes – List the unit prefixes in their appropriate decimal position



Dimensional Analysis

Convert the following:

$$20 \text{ mi hr}^{-1} \rightarrow \text{m s}^{-1}$$

$$\frac{20 \text{ mi}}{1 \text{ hr}} \times \frac{1609 \text{ m}}{1 \text{ mi}} \times \frac{1 \text{ hr}}{60 \text{ min}} \times \frac{1 \text{ min}}{60 \text{ s}} = 8.9 \frac{\text{m}}{\text{s}} = 8.9 \text{ m s}^{-1}$$

$$0.0007 \text{ km}^2 \rightarrow \text{m}^2$$

$$0.0007 \text{ km}^2 \times \frac{1000 \text{ m}}{1 \text{ km}} \times \frac{1000 \text{ m}}{1 \text{ km}} = 700 \text{ m}^2 \quad \text{or} \quad 0.0007 \text{ km}^2 \times \left(\frac{1000 \text{ m}}{1 \text{ km}}\right)^2 = 700 \text{ m}^2$$

Determine the units for Q:

$$Q = mc \Delta T$$

$$Q = (\text{kg})(\text{J kg}^{-1} \text{K}^{-1})(\text{K}) = \frac{(\text{kg})(\text{J})(\text{K})}{\text{kg K}} = \text{J}$$

m (mass)	kg
c (specific heat)	$\text{J kg}^{-1} \text{K}^{-1}$
ΔT (change in temp)	K

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Pat Roby



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