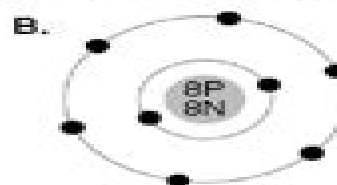
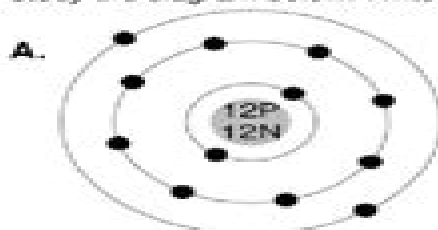


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IPS Unit 9 – Chemical Bonds

Types of Bonds Worksheet

Directions: Study the diagram below. Write your answers to the questions in the spaces provided.



1. If atom A loses electrons to atom B.
 - a. how many electrons will atom A lose? _____
 - b. how many electrons will atom B gain? _____
 - c. what will be the charge of atom A? _____
 - d. what will be the charge of atom B? _____
 - e. what will be the total charge of the compound formed?

 - f. what type of bond will form?

2. Explain why an element's charge is related to the group on the periodic table to which it belongs.

Directions: Complete the table comparing ionic compounds and covalent compounds.

Characteristic	Ionic compounds	Covalent compounds
How the compound is formed		
Smallest particle		
Usual state at room temperature		

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Amar Nath Bhutani, Saurabh Joshi



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that credit was assigned Gilbert Lewis for example could be reclusive and resentful and his enmity with Walther Nernst may have cost him the Nobel Prize Irving Langmuir gregarious and charming rediscovered Lewis's theory of the chemical bond and received much of the credit for it Langmuir's personality smoothed his path to the Nobel Prize over Lewis Coffey deals with moral and societal issues as well These same scientists were the first to be seen by their countries as military assets Fritz Haber dubbed the father of chemical warfare pioneered the use of poison gas in World War I vividly described and Glenn Seaborg and Harold Urey were leaders in World War II's Manhattan Project Urey and Linus Pauling worked for nuclear disarmament after the war Science was not always fair and many were excluded The Nazis pushed Jewish scientists like Haber from their posts in the 1930s Anti-Semitism was also a force in American chemistry and few women were allowed in Pauling for example used his influence to cut off the funding and block the publications of his rival Dorothy Wrinch Cathedrals of Science paints a colorful portrait of the building of modern chemistry from the late 19th to the mid 20th century

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