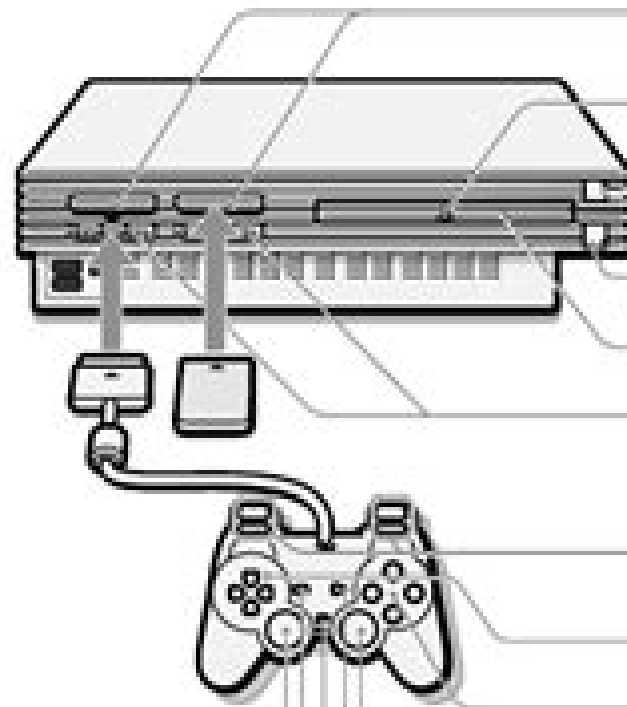


SONY

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PlayStation 2



Ps2 User Manual

**United States. National Bureau of
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Game Console Hacking Joe Grand, Albert Yarusso, 2004-11-12 The worldwide video game console market surpassed 10 billion in 2003 Current sales of new consoles is consolidated around 3 major companies and their proprietary platforms Nintendo Sony and Microsoft In addition there is an enormous installed retro gaming base of Ataria and Sega console enthusiasts This book written by a team led by Joe Grand author of Hardware Hacking Have Fun While Voiding Your Warranty provides hard core gamers with they keys to the kingdom specific instructions on how to crack into their console and make it do things it was never designed to do By definition video console game players like to have fun Most of them are addicted to the adrenaline rush associated with winning and even more so when the winning involves beating the system by discovering the multitude of cheats built into most video games Now they can have the ultimate adrenaline rush actually messing around with the soul of the machine and configuring it to behave exactly as the command This book builds on the motto of Have Fun While Voiding Your Warranty and will appeal to the community of hardware geeks who associate unscrewing the back of their video console with para jumping into the perfect storm Providing a reliable field tested guide to hacking all of the most popular video gaming consoles Written by some of the most knowledgeable and recognizable names in the hardware hacking community Game Console Hacking is the first book on the market to show game enthusiasts self described hardware geeks how to disassemble reconfigure customize and re purpose their Atari Sega Nintendo Playstation and Xbox systems *Technical Report* ,2003 Official Gazette of the United States Patent and Trademark Office ,2003

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 mL H_2SO_4 4 mL in the 125 mL flask and stir

continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0 ml of nitric acid and 4.0 ml of sulfuric acid and will then react with 4.5 ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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