

Physical Science – 2nd Semester – Final Exam Study Guide – (2012–2013)

1. Physical Science Basics

- What tool(s) would you want to use to find the:
 - Mass of an object? triple beam balance
 - Volume of an object? ruler or water displacement method
 - Length of an object? Ruler
 - Density of an object? graduated cylinder
- What is the formula for calculating density? mass/volume
 - What is the density of a cube that is 33 g and has a volume of 3 cm³? 33/3cm³
 - Compare the density of solid H₂O to liquid H₂O, and explain what observable effect this has.
solid H₂O less dense than liquid H₂O

2. Chemistry Basics

- Element (def.) substance made up of one kind of atom
- Atom (def.) basic building block of matter
- Compound (def.) a molecule that contains atoms or molecules of two or more different elements
- Molecule (def.) a combination of two or more of the same or different atoms that are bonded together

3. Model of the Atomic Evolution of the Atomic Theory

Atoms are so small that many scientists throughout history have created models to describe them.

Why have atomic theories changed over time? Better technology came around.

Scientist	Year	Major Discovery/ Contribution	Name of atomic model
John Dalton	1808	Each element is made up of small, indivisible particles called atoms. Different elements have atoms of different masses.	Billiard Ball Model
J.J. Thomson	1897	Negatively charged electron	Plum Pudding Model
Niels Bohr	1913	Electrons move <u>fixed orbits</u> around the atom's nucleus in specific layers, or <u>electron shells</u> . Atoms absorb or give off energy when the electrons move from one shell (<u>energy level</u>) to another.	Bohr Model
Ernest Rutherford	1909	1909 /Based on idea that positive reflects positive/ atom mostly empty space; electrons orbit randomly around positively charged nucleus	Planetary Model

4. Periodic Table

- The current periodic table is arranged by increasing atomic number.
- Reading the Periodic Table
 - An element's properties can be predicted from its location in the periodic table.
 - The atomic number represents the number of protons.
 - The atomic mass represents the number of neutrons and protons.

Physical Science Semester 2 Exam Study Guide

Lucas Blade Hebert



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Directory of Distance Learning Opportunities Modoc Press, Inc., 2003-02-28 This book provides an overview of current K 12 courses and programs offered in the United States as correspondence study or via such electronic delivery systems as satellite cable or the Internet The Directory includes over 6 000 courses offered by 154 institutions or distance learning consortium members Following an introduction that describes existing practices and delivery methods the Directory offers three indexes Subject Index of Courses Offered by Level Course Level Index Geographic Index All information was supplied by the institutions Entries include current contact information a description of the institution and the courses offered grade level and admission information tuition and fee information enrollment periods delivery information equipment requirements credit and grading information library services and accreditation *The Hidden Curriculum - Faculty Made Tests in Science* Sheila Tobias, 1997 Medical Laboratory Technician Certification Study Guide 2025-2026 Lucas Blade Hebert, Master Laboratory Science Ace Your Certification Ready to excel in your MLT or MLS certification exam This comprehensive review guide provides everything you need to succeed from fundamental concepts to advanced clinical applications 700 Practice Questions Test your knowledge across all laboratory disciplines with detailed explanations that reinforce learning 26 Complete Chapters Master Hematology Chemistry Microbiology Immunology Blood Banking and more with systematic easy to follow content Real Clinical Cases Apply your knowledge through authentic laboratory scenarios that mirror actual practice High Yield Review Focus your study time on the most frequently tested concepts with targeted review sections Test Taking Strategies Learn proven techniques to maximize your performance on examination day From basic laboratory principles to complex diagnostic challenges this guide builds the knowledge and confidence you need for certification success Perfect for MLT and MLS candidates laboratory science students and practicing professionals seeking continuing education Transform your preparation Achieve your certification goals Resources in Education , 1995-07 **Research in Education** , 1974 **Vocational Education Bulletin** United States. Division of Vocational Education, 1961 *Peterson's Graduate Programs in the Physical Sciences, Mathematics, Agricultural Sciences, the Environment & Natural Resources* 2012 Peterson's, 2011-12-30 Graduate Programs in the Physical Sciences Mathematics Agricultural Sciences the Environment Natural Resources 2012 contains more than 2 900 graduate programs in 59 disciplines including agriculture and food sciences astronomy and astrophysics chemistry physics mathematics environmental sciences and management natural resources marine sciences and more This guide is part of Peterson's six volume Annual Guides to Graduate Study the only annually updated reference work of its kind provides wide ranging information on the graduate and professional programs offered by U S accredited colleges and universities in the United States and throughout the world Informative data profiles for more than 2 900 graduate programs in 59 disciplines including facts and figures on accreditation degree requirements application deadlines and contact information financial support faculty and student body profiles Two page in depth

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Peterson's Graduate Programs in the Physical Sciences 2011 Peterson's,2011-05-01 Peterson's Graduate Programs in the Physical Sciences contains a wealth of information on colleges and universities that offer graduate work in Astronomy and Astrophysics Chemistry Geosciences Marine Sciences and Oceanography Meteorology and Atmospheric Sciences and Physics The institutions listed include those in the United States Canada and abroad that are accredited by U S accrediting bodies Up to date information collected through Peterson's Annual Survey of Graduate and Professional Institutions provides valuable information on degree offerings professional accreditation jointly offered degrees part time and evening weekend programs postbaccalaureate distance degrees faculty students degree requirements entrance requirements expenses financial support faculty research and unit head and application contact information As an added bonus readers will find a helpful See Close Up link to in depth program descriptions written by some of these institutions These Close Ups offer detailed information about the physical sciences program faculty members and their research and links to the program or department's Web site In addition there are valuable articles on financial assistance and support at the graduate level and the graduate admissions process with special advice for international and minority students Another article discusses important facts about accreditation and provides a current list of accrediting agencies

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