



Atomic Structure and Nuclear Chemistry Study Guide

Topics Covered:

Subatomic particles and the nucleus
Isotopes and Ions
Atomic mass and atomic number
Average atomic mass and isotopic abundance
Nuclear stability
Ionizing radiation
Alpha decay and beta decay
Transmutation equations
Fission vs. fusion
Mass spectroscopy
Half-life

Practice Questions: [Click here for an answer key](#)

1. What does it mean for an atom to be neutral? What does it mean for different subatomic particles play here?
2. What is the relationship between the atomic number/atomic mass and

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Joel J. Mintzes



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