

Introduction to Chemistry: Chapter 1 Outline

1.1 The Scope of Chemistry

A. What is Chemistry?

1. **Matter**- anything that has mass and takes up space
 - You don't have to be able to see something for it to be considered as matter.
 - Light is an example of something that does **NOT** have matter
2. **Chemistry**- Study of the composition of matter and the changes matter undergoes
 - Chemistry affects all aspects of life and most natural events because all living and nonliving things are made of matter
 - Chemistry is known as a central science because it is useful to understanding other sciences.
 - Ozone (O₃) is a molecule that absorbs ultraviolet light
 - Chlorofluorocarbons-
3. **Mass** - a measurement that reflects the amount of matter
4. **Weight**- is a measurement that reflects the pull of gravity on a given mass
 - Mass and weight are different because your weight changes while on the moon

B. Areas of Study

- Five traditional areas of study are organic chemistry, inorganic chemistry, biochemistry, analytical chemistry, and physical chemistry.
 - Most chemicals found in organism contain carbon.
 1. **Organic Chemistry**- study of all chemicals containing carbon
 2. **Inorganic Chemistry**- study of chemicals not containing carbon.
 - Many inorganic chemicals are found in non-living things.
 3. **Biochemistry**- Study of processes that take place in living organisms
 4. **Analytical Chemistry**- Area of study that focuses on the composition of matter (Analyze)
 5. **Physical Chemistry**- Area that deals with mechanism, rate and energy transfer that occurs when matter undergoes a change. Study of molecular shape and function.
 6. **Pure Chemistry**- Pursuit of Chemical knowledge for its own use.
 7. **Applied Chemistry**- research directed toward a practical goal or application.
 - An example of applied could be a chemist looking to find a cure for an illness.
 - Chemist don't expect for an immediate use for this knowledge, but most chemist research that's designed to answer a specific question.
 - Pure and Applied chemistry are linked. Pure research can lead to an application while an application can exist before the research is done to explain how it works.
- Two Types of Data:
8. **Qualitative**- data that is descriptive, non-numeric information
 9. **Quantitative**- data is a numeric measurement with a unit associated with it.

C. Big Ideas in Chemistry

Big Ideas:

- Chemistry as the central science- Chemistry overlaps with all of the other sciences.

Outline Of Chemistry A Summary For First Year Students

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