Name:	Score :	Date:	

## Practicing Protons, Neutrons, and Electrons

1. Fill in the element name and the number of protons and neutrons for the given symbols.

<sup>4</sup><sub>2</sub>He

56 Fe

27AI

Element name: # of protons:

# of neutrons:

Element name: # of protons: \_\_\_\_

# of neutrons:

Element name: # of protons: # of neutrons:

 $^{40}_{20}$  Ca

56 Ne

12 Ca

Element name: # of protons:

# of protons: # of neutrons:

Element name: Element name:

# of protons: # of neutrons:

 $^{19}F$ 

# of neutrons:

35 CI Element name: 23 Na

Element name: # of protons:

# of neutrons:

# of protons: # of neutrons:

Element name: # of protons: # of neutrons:

Using the periodic table, fill in the missing information for the given elements.

Element	Atomic Number	Atomic Mass	Number of protons	Number of neutrons	Number of electrons
S	16		16		
	17	35			17
	18		18	22	
	19	39			19
Ca	20		20		

# **Protons Neutrons Electrons Practice Answers**

**Julia Schneider** 

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