

# Stoichiometry of a Precipitation Reaction – Lab Report Assistant

## Exercise 1: Stoichiometry and a Precipitation Reaction

**Data Table 1.** Stoichiometry Values

|                                                            |            |
|------------------------------------------------------------|------------|
| Initial: $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ (g)     | 1.00 g     |
| Initial: $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ (moles) | .00680 mol |
| Initial: $\text{CaCl}_2$ (moles)                           | .00680 mol |
| Initial: $\text{Na}_2\text{CO}_3$ (moles)                  | .00680 g   |
| Initial: $\text{Na}_2\text{CO}_3$ (g)                      | .721 g     |
| Theoretical: $\text{CaCO}_3$ (g)                           | 0.681      |
| Mass of Filter paper (g)                                   | 1.00 g     |
| Mass of Filter Paper + $\text{CaCO}_3$ (g)                 | 1.42 g     |
| Actual: $\text{CaCO}_3$ (g)                                | .42 g      |
| % Yield:                                                   | 61.69%     |

### Questions:

- A perfect percent yield would be 100%. Based on your results, describe your degree of accuracy and suggest possible sources of error.
- What impact would adding twice as much  $\text{Na}_2\text{CO}_3$  than required for stoichiometric quantities have on the quantity of product produced?
- Determine the quantity (g) of pure  $\text{CaCl}_2$  in 7.5 g of  $\text{CaCl}_2 \cdot 9\text{H}_2\text{O}$ .
- Determine the quantity (g) of pure  $\text{MgSO}_4$  in 2.4 g of  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ .
- Conservation of mass was discussed in the background. Describe how conservation of mass (actual, not theoretical) could be checked in the experiment performed.

# Precipitation Reactions Lab Report Answers

**V. K. Sally, Chhaya Srivastava, Goyal  
Brothers Prakashan**



## Precipitation Reactions Lab Report Answers:

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*Energy Research Abstracts* , 1993      **Metals Review** , 1958      **Ecosystem Biogeochemistry** Christopher S. Cronan, 2017-10-05 This textbook presents a comprehensive process oriented approach to biogeochemistry that is intended to appeal to readers who want to go beyond a general exposure to topics in biogeochemistry and instead are seeking a holistic understanding of the interplay of biotic and environmental drivers in the cycling of elements in forested watersheds The book is organized around a core set of ecosystem processes and attributes that collectively help to generate the whole system structure and function of a terrestrial ecosystem In the first nine chapters a conceptual framework is developed based on distinct soil microbial plant atmospheric hydrologic and geochemical processes that are integrated in the element cycling behavior of watershed ecosystems With that conceptual foundation in place students then proceed to the final three chapters where they are challenged to think critically about integrated element cycling patterns roles for biogeochemical models the likely impacts of disturbance stress and management on watershed biogeochemistry and linkages among patterns and processes in watersheds experiencing novel environmental changes Included with the text are figures tables of comparative

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This book delves into Precipitation Reactions Lab Report Answers. Precipitation Reactions Lab Report Answers is a crucial topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Precipitation Reactions Lab Report Answers, encompassing both the fundamentals and more intricate discussions.

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